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literature on non-equilibrium thermodynamics and its relationship to evolutionary processes. 2 Non-Equilibrium Thermodynamics: How It Works 2.1 The 2nd Law The second law of thermodynamics asserts that if a spontaneous reaction occurs, the reaction moves towards an irreversible state of equilibrium and in the process, becomes increasingly random or disordered. It is this increasing disorder or

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Non equilibrium thermodynamics in multiphase flows

Non equilibrium thermodynamics in multiphase flows 1, Chapter 2 Microscopic Reversibility The Principle of Microscopic Reversibility was formulated by Richard Tolman [14] who stated that, at equilibrium, any molecular process and the reverse of that process will be taking place on the average at the same rate.

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